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400

8 Group III-V NITRIDE-BASED ULTRAVIOLET LEDs AND LASER DIODE

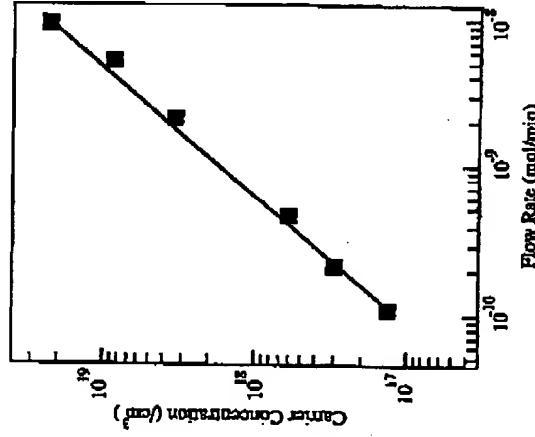


Fig. 7. Carrier concentrations of silicon-doped gallium nitride films as a function of the flow rate of silane (SiH_4). Reproduced from Nakamura *et al.* (1992c) with the permission of the Japanese Journal of Applied Physics.

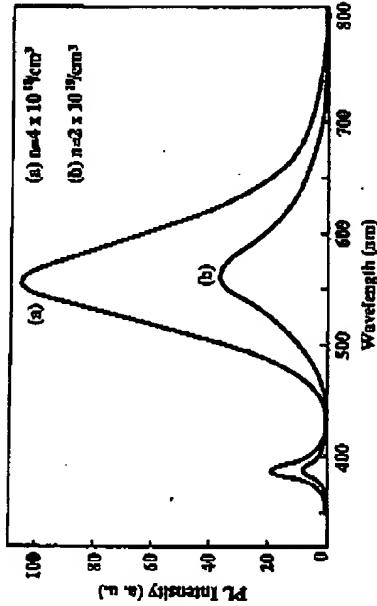


Fig. 8. Photoluminescence (PL) spectra of silicon-doped gallium nitride (GaN) films grown with GaN buffer layers under the same growth conditions except for the flow rate of silane (SiH_4). The flow rates for SiH_4 were (a) $2 \text{ nmol}/\text{min}$ and (b) $10 \text{ nmol}/\text{min}$. The carrier concentrations were (a) $4 \times 10^{18} \text{ cm}^{-3}$ and (b) $2 \times 10^{18} \text{ cm}^{-3}$. Reproduced from Nakamura *et al.* (1992c) with the permission of the Japanese Journal of Applied Physics.

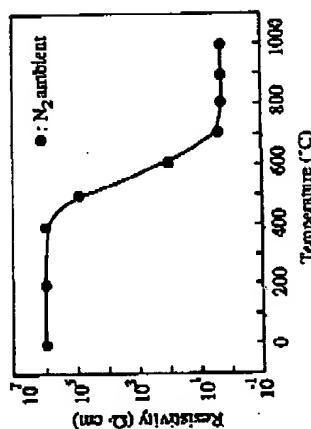


Fig. 9. Resistivity of as-grown magnesium-doped gallium nitride films as a function of annealing temperature. N_2 : nitrogen. Reprinted from Nakamura *et al.* (1992c) with the permission of the Japanese Journal of Applied Physics.

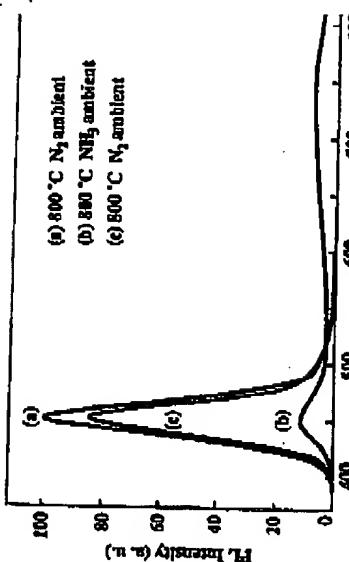


Fig. 10. Photoluminescence (PL) spectra of as-grown magnesium-doped gallium nitride films that were annealed at different temperatures: (a) room temperature, (b) 700°C, and (c) 800°C. R.T., room temperature. Reprinted from Nakamura et al. (1992a) with the permission of the Japanese Journal of Applied Physics.

temperatures above 600°C. In the case of N₂ ambient thermal annealing at temperatures between room temperature and 1000°C, the low-resistivity p-type GaN films showed no change in resistivity, which was almost constant between 2 and 8 Ω·cm, as shown in Fig. 11.

Figure 12(a) shows the PL spectrum of 800°C N₂ ambient thermal-annealed GaN film, Fig. 12(b) shows the film after NH₃ ambient thermal-annealing at 800°C for the sample in Fig. 12(a), and Fig. 12(c) shows the

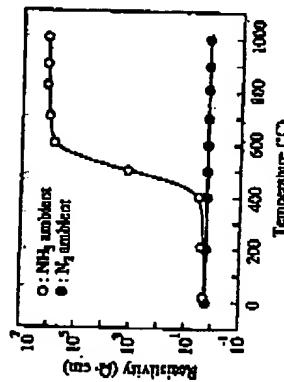


Fig. 11. The resistivity change in N₂ ambient thermal-annealed low-resistivity magnesium-doped gallium nitride films as a function of annealing temperature. The ambient gases, ammonia (NH₃) and nitrogen (N₂), were used for thermal annealing. Reprinted from Nakamura et al. (1992a) with the permission of the Japanese Journal of Applied Physics.

film after N₂ ambient thermal annealing at 800°C for the sample in Fig. 12(b). Before NH₃ ambient thermal annealing, the intensity emission is strong, and the broad DL emission is not observed (Fig. 12(a)). After NH₃ ambient thermal annealing at 800°C for the sample in Fig. 12(a), the intensity of the blue emission becomes weak, and the DL emission around 750 nm appears (Fig. 12(b)). The PL spectra after NH₃ ambient thermal annealing at 800°C. These changes were found to be reversible with a change in the ambient gas from NH₃ to N₂, as is the case with the resistivity.

These results indicate that atomic hydrogen produced by NH₃ at temperatures above 400°C is related to the acceptor complex mechanism. A hydrogenation process whereby acceptor-H complexes are formed in p-type GaN films was proposed. The formation of acceptor-H neutral complexes causes acceptor compensation, and the hydrogen atoms occur at the surface of GaN film. Dangling bonds exist mainly at the surface, and the atomic hydrogen diffuses into the GaN films because the number of hydrogen atoms at the surface and the size of hydrogen atoms is very small.

atomic hydrogen, produced by dissociation of NH₃ at temperature 400°C, diffuses into p-type GaN films. Second, the formation of a neutral complex, that is, Mg-H complexes in GaN films ex-